

CHEMISTRY SINGLE REPLACEMENT REACTION WORKSHEET

Practice Reactions:

1. $\text{Ag} + \text{KNO}_3 \rightarrow$
2. $\text{Zn} + \text{AgNO}_3 \rightarrow$
3. $\text{Al} + \text{H}_2\text{SO}_4 \rightarrow$
4. $\text{Cl}_2 + \text{KI} \rightarrow$
5. $\text{Li} + \text{H}_2\text{O} \rightarrow$
6. $\text{Cu} + \text{FeSO}_4 \rightarrow$
7. $\text{Na} + \text{H}_2\text{O} \rightarrow$
8. $\text{Fe} + \text{Pb}(\text{NO}_3)_2 \rightarrow$
9. $\text{Cu} + \text{H}_2\text{O} \rightarrow$
10. $\text{Cu} + \text{Al}_2(\text{SO}_4) \rightarrow$
11. $\text{Al} + \text{Pb}(\text{NO}_3)_2 \rightarrow$
12. $\text{Cl}_2 + \text{NaI} \rightarrow$
13. $\text{Fe} + \text{AgC}_2\text{H}_3\text{O}_2 \rightarrow$
14. $\text{Al} + \text{CuCl}_2 \rightarrow$
15. $\text{Br}_2 + \text{CaI}_2 \rightarrow$
16. $\text{Al} + \text{HCl} \rightarrow$
17. $\text{Mg} + \text{HCl} \rightarrow$
18. $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow$
19. $\text{Fe} + \text{CuSO}_4 \rightarrow$
20. $\text{Cl}_2 + \text{MgI}_2 \rightarrow$